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Diastereoselective Radical Cyclization Using a Chiral α -Methyl- α,β -unsaturated Ester: Controlling the Stereochemistry at both the α - and β -Positions

Mayumi Nishida,* and Manami Nobuta

Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo 060, Japan

Kazuyo Nakaoka

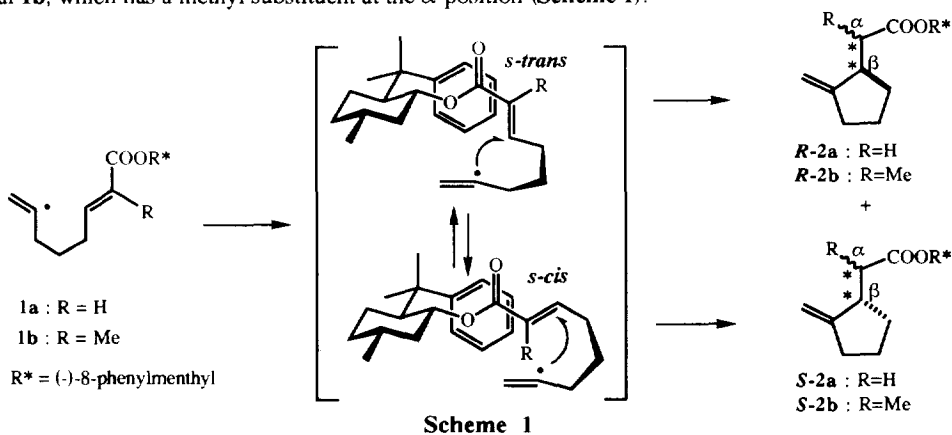
Center for Instrumental Analysis, Hokkaido University, Sapporo 060, Japan

Atsushi Nishida,* and Norio Kawahara

Hokkaido Institute of Pharmaceutical Sciences, Katsuraoka 7-1, Otaru 047-02, Japan

Abstract: Diastereoselective radical cyclizations of (-)-8-phenylmenthyl 2-methyl-2-octene-7-ynoate (**9**) and (-)-8-phenylmenthyl 7-iodo-2-methyl-2,7-octadienoate (**10**) were investigated. The best results were obtained in the cyclization of **10** at -98°C . A mixture of four diastereomers was obtained at a yield of 72% and with a diastereoselectivity of 78:12:9:1. The main product was (*S*)-2-[(*R*)-2-methylene-cyclopentyl]propionate (**3a**). The diastereoselectivities at the α - and β -positions were 90:10 and 79:21, respectively.

We recently reported β -diastereoselective radical cyclizations of **1a** with α,β -unsaturated (-)-8-phenylmenthyl ester^{1, 2} as a chiral radical acceptor.³ High selectivity was observed when the reaction was carried out in the presence of Lewis acids, which gave an unsaturated ester with an *s-trans* conformation. We report here our efforts to control the stereochemistry at both the α - and β -position in the cyclization of alkenyl radical **1b**, which has a methyl substituent at the α -position (Scheme 1).



Four possible diastereomers, **3a-d**, can be obtained in the cyclization of **1b** (Figure 1). Hence, before the radical reactions were investigated, these diastereomers were synthesized by alternative routes.

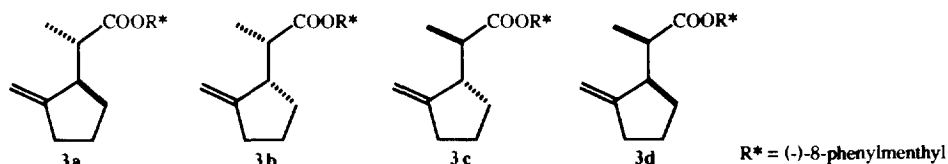
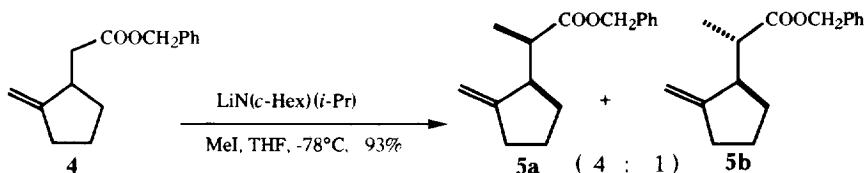
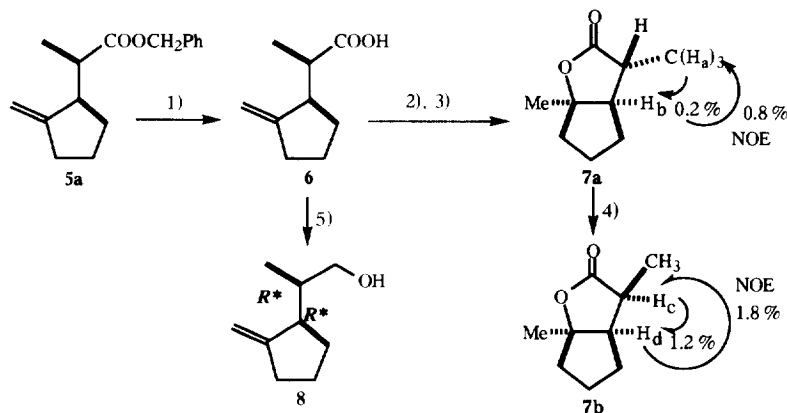


Figure 1 Four possible diastereomers



Scheme 2

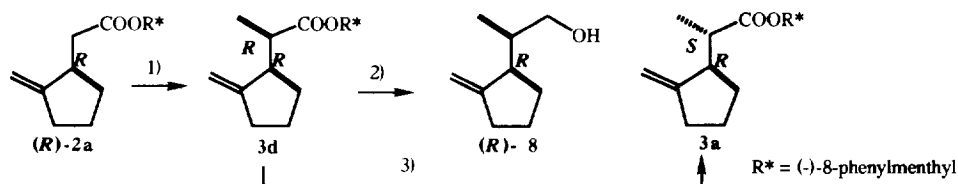
Methylation of racemic ester **4** gave a mixture of **5a** and **5b** at a yield of 93% and with a diastereoselectivity of 4 : 1 (**Scheme 2**). The stereochemistry of the main product **5a** was determined by NOE experiments after conversion to **7a** and then to diastereomer **7b** (**Scheme 3**). In NOE studies of **7a**, irradiation of H_a resulted in a 0.2% NOE to H_b and irradiation of H_b resulted in a 0.8% NOE to H_a . Furthermore, irradiation of H_c in **7b**, resulted in a 1.2% NOE to H_d and irradiation of H_d resulted in a 1.8% NOE to H_c . Thus, the relative stereochemistry of **5a** was determined as shown.



1) 3*N* NaOH, MeOH, THF, 84%, 2) I₂, NaHCO₃, THF, 0°C, 89%, 3) *n*-Bu₃SnH, AIBN, benzene, reflux, 92%, 4) LDA, THF, then isopropanol, -78°C, 86%, 5) LAH, ether, 0°C, 91%.

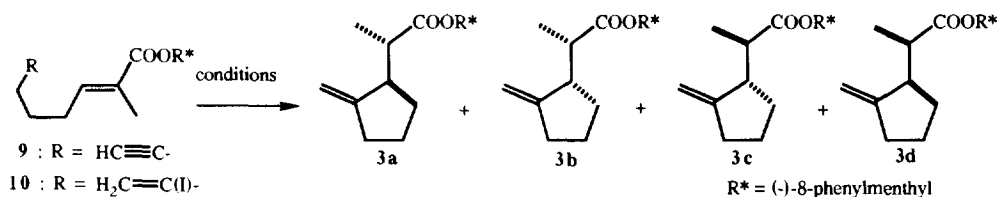
Scheme 3

Methylation of (*R*)-**2a**^{3a} and subsequent reduction gave the optically active alcohol (*R*)-**8** as a single diastereomer (**Scheme 4**)⁴. Since the relative configuration of (*R*)-**8** was comparable to that of racemic alcohol **8**, which was prepared from **6**, **3d** was determined to be (-)-8-phenylmethyl (*R*)-2-[(*R*)-2-methylenecyclopentyl]propionate. Under basic conditions, **3d** gave **3a**. The other diastereomers, **3b** and **3c**, were synthesized by the same method using (*S*)-**2a**.^{3a,4}



1) Lithium *N*-isopropylcyclohexylamide, MeI, THF, -78°C, 97%, 2) LAH, ether, 0°C, 91%, 3) Lithium *N*-isopropylcyclohexylamide, then isopropanol, THF, -78°C, 86%.

Scheme 4

Table 1 Intramolecular radical cyclizations of **9** and **10**

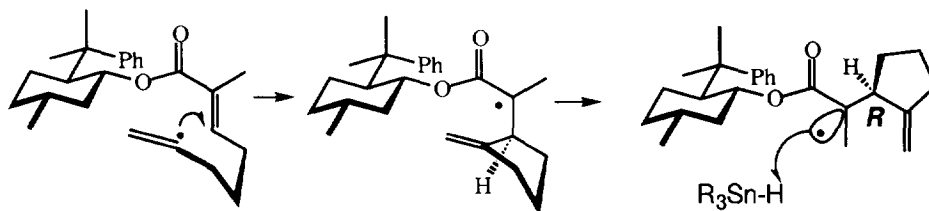
conditions : **A)** *n*-Bu₃SnH (1.5 eq.), AIBN (cat.), benzene.

B) *n*-Bu₃SnH (1.5 eq.), Et₃B (1.05 eq.), toluene.

run	substrate 9 or 10	conditions	temp. (°C)	yield (%)	ratio ^{a)} 3a : 3b : 3c : 3d	selectivity	
						α (3a+3b) : (3c+3d)	β (3a+3d) : (3b+3c)
1	9	A	80	82 ^{b)}	66 : 18 : 9 : 7	84 : 16	73 : 27
2	9	B	0	77 ^{b)}	67 : 18 : 9 : 6	85 : 15	73 : 27
3	10	A	80	76	60 : 20 : 10 : 10	80 : 20	70 : 30
4	10	B	0	73 ^{c)}	71 : 12 : 11 : 6	83 : 17	77 : 23
5	10	B	-98	72 ^{c)}	78 : 12 : 9 : 1	90 : 10	79 : 21

a) The product ratio was determined by ¹H-NMR. b) A tributylstannyl group in the initial products was removed by a treatment with BF₃•OEt₂. c) (-)-8-Phenylmenthyl 2-methyl-2,7-octadienoate was also obtained at a yield of 7%.

In the cyclization of **9** under thermal condition **A**, **3a** was obtained as a major product. The diastereoselectivities at the α - and β -positions were 84:16 and 73:27, respectively (**Table 1**, run 1). Cyclization at 0 °C (condition **B**)⁵ did not improve the diastereoselectivity (run 2).⁶ The reactivity of stannyl radical toward an acetylenic moiety was low below 0°C.^{2a} We then used **10** as a substrate, which easily produces an alkenyl radical even at -98°C. The cyclization of **10** at -98°C gave a modest increase in the diastereoselectivity at both the α - and β -positions. A mixture of the cyclized products was obtained in 72% yield. The selectivity at the α -position was 90 : 10 and that at the β -position was 79 : 21.

**Scheme 5** Transition-state model

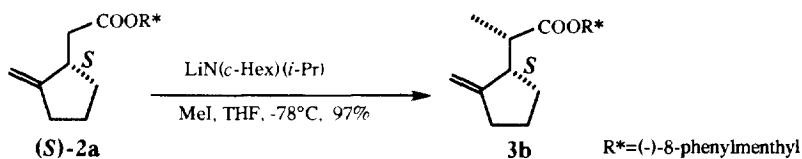
The stereochemical outcome observed in these radical reactions suggests a working hypothesis for the transition states (**Scheme 5**). The phenyl group shields a π -face of the alkene, and the α -methyl- α,β -unsaturated ester exists in an *s-trans* conformation. The alkenyl radical attacked the β -position of the α -methyl- α,β -unsaturated ester from a less-hindered face of the alkene. The resulting α -radical is planar, since it is

conjugated with carboxylic ester and appears to favor the conformation with the larger cyclopentyl group *anti* to the alkoxy group.^{2b} Hydrogen abstraction occurs predominantly from the *Si*-face of the α -radical.

In conclusion, we have reported the intramolecular cyclization of alkenyl radicals using α -methyl- α,β -unsaturated (-)-8-phenylmenthyl ester as a chiral radical acceptor, through which we can control the diastereoselectivities at the α - and β -position. A moderate temperature dependence was observed and the highest diastereoselectivity was observed in the reaction at -98°C.

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- The stereochemistry of the methylation of (*R*)-**2a** did not appear to be controlled by (-)-8-phenylmenthyl as a chiral auxiliary. Instead, it was controlled by the adjacent stereogenic carbon. The methylation of (*S*)-**2a** proceeded stereoselectively.



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- In the reaction of **11**, the presence of Lewis acids was essential to achieve high diastereoselectivity.^{3a} However, the cyclization of **9** in the presence of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (2 or 32 eq) at 0°C was inefficient.

